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AUTOMATED QUANTITATIVE GAS-LIQUID CHROMATOGRAPHY OF INTACT LIPIDS of associated below the spectrum of

# **T. PREPARATION AND CALIBRATION OF THE COLUMN**

# P. MARES\*, E. TVRZICKÁ and V. TAMCHYNA

Gastroenterological Research Unit I, Faculty of Medicine, Charles University, Prague (Czechosłovakia) je popozna – pagrupa i inalizijske politik in preznanda – polit angal pilagi menyerakan ka ชาวไฟไซล์เลย (Son) (เฮอ์) ไซเลี้ยนไซเลีย (Son) (เลย (เฮอ) ค

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# SUMMARY AND SECRETIC BELLEVILLE sdal thum naigh. The homels was in this solicity of the

.... A method for quantitative gas chromatographic determination of plasma lipids (free cholesterol, cholesteryl esters and triglycerides) in the low concentration range is described. This method permits a determination of not only the lipid classes mentioned above, but also their fractions according to molecular weight, down to 10 ng, without previous derivatization. Special attention was devoted to the preparation of columns with high efficiency and minimal losses of the test substances. The best results were obtained with a glass column  $0.5$  m  $\times$  2.0 mm I.D., packed with 1% OV-1 on Gas-Chrom Q (100-120 mesh). The processing of results is fully automated, using an MDS-2400 computer and includes the calculation of a non-linear calibration plot for each substance analyzed, accuracy control of the measured values, tabulation of the  $f_{\omega r}$  values and the calculation for analyses of biological samples. For the calibration, the pure substances were used at 15 concentrations within a range of 10-1000 ng. The coefficient of variation calculated from 20 duplicate measurements of the calibration mixture did not exceed 5% for any component in the interval from 10 to 100 ng or 3% within range from 100 to 1000 ng. Professional and professional

# INTRODUCTION SECTION

<u>ISTORIA (LITERATURIA ANGLETIKA</u>

The Goldsman planet (THE moterman building () of a set of Direct gas chromatographic (GC) analysis of plasma lipids was first described by Kuksis et al. [1] more than ten years ago. Since then, great advances have been made not only in the equipment, but also in the field of computing technique. In 1975 it was again Kuksis et al. [2] who published their experiences with fully automated GC estimation of the plasma-lipid profile. In the development of GC equipment, supports and stationary phases, as well as in the quality of standards, great progress has been made during the last decade.

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**At present, the detection limit of high-molecular substances, such as lipids, is**  lower than 50 ng. However, quantitative analysis of such low concentrations is not carried out as a routine procedure. The main reason is the non-linear weight correction factor  $(f_w)$  vs. concentration plot for the substances analyzed [3], which was observed e.g. by Bezard and Bugaut [4, 5], even at higher concentrations. In a higher concentration range (of the order of thousands of ng), the **fw value is independent of the amount of lipid analyzed [S]** ; **the concentration**  limit depends on the quality of the column and on the whole chromatographic system. With biological samples, the determination of much lower concentrations is necessary. This paper describes a computerized, fully automated pro**cedure for the GC determination of neutral lipids down to 10 ng** bf each **domponent. The sample preparation, accuracy and precision of the analysis of biological samples will be the object of the following communicatioris:** 

#### **MATERIALS**

**Triglycerides with carbon numbers .48, 50, 52, 54 and: 60, purity 99%**  (tripalmitin, rac-glyceryl-1,3-palmitate-2-stearate, rac-glyceryl-1,3-stearate-2palmitate, tristearin and triarachidin) were obtained from Supelco (Bellefonte, Pa., U.S.A.) and Sigma (St. Louis, Mo., U.S.A.), respectively. Cholesteryl **esters with carbon numbers 31,41,43,45 and 47 (cholesteryl butyrate, cholesteryl myristate, cbolesteryl paImitate, cbolesteryl stearate and cholesteryl arachidate) and free cholesterol were obtained from Applied Science Labs.**  (State College, Pa., U.S.A.). The OV-1 stationary phase and the Gas-Chrom **& support (100-120 mesh), were also supplied by Apphed Science Labs. Isooct&e (analytical grade) was provided by International Enzymes (Windsor,**  Great Britain). Chloroform, methanol, acetone and toluene, all analytical **grade, were obtained from Lachema (Bmo, Czechoslovakia). Helium of 99.99% purity was supplied by Messer (Griesheim, G.F.R.) -and trimethylchlorosilan was provided by Merck (Darmstadt, G.F.R.). \_:**  *.* 

#### Apparatus and operating conditions

All analyses were performed on a Perkin-Elmer F 30 gas chromatograph (Norwalk, Conn., U.S.A.) equipped with a dual column system with flameionization detection. Samples were injected by means of a Perkin-Elmer PS **4950 liquid autosampler into a l/4 in. glass-lined injector thermostated to 300".**  Oven temperature was programmed as follows: initial temperature 180°, **programme rate 5"/min, final temperature 350", detector temperature 350°.**  Helium flow-rate was 100 ml/min. The gas chromatograph was combined with a Perkin-Elmer Model 56 recorder, a Perkin-Elmer M-2 calculating integrator and a Teletype 33 ASR-FR (Teleprint, Frankfurt, G.F.R.) equipped with **paper-tape puncher and reader. Data wore processed by- a' computer** MDS-2400 (32 K Byte core, 2 magnetic tape units, line printer, paper-tape reader) **using a special programme.**  $\alpha$  is the set of the second construction policy of the second  $\frac{1}{2}$  , the contract of the second contract of the contract of  $\frac{1}{2}$  ,  $\frac{1}{2}$ 

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#### **METHODS**

#### Preparation of the column

**. . A glass .column. 9.5 m X** .2.0 **mm ID.- was rinsed with 150~ml volumes of chloroform, methanol and &&one end dried. Then the column wasfilled with a 10% solution of trimethylchlorosilane in anhydrous toluene, allowed to stand for 10 min and rinsed with toluene. The column was then treated for 5 min with methanol and rinsed with the same solvent to the neutial reaction. After drying in an oven, the column was packed immediately with 1% OV-1 on**  Gas-Chrom Q (100 - 120 mesh), using the combined effect of suction and vibration. The packing was prepared by the evaporation technique and pre**stabilized in a stream of helium, initially at a temperature programme SO-**350°, 1°/min, then for 3<sup>h</sup> at 350°. After testing the separation ability (by **means of a staudard mixture of triglycerides and cholesteryl esters), the column was further stabilired by 5 injections of 10,000~ng amounts of triolein. The reproducibility of the recovery was then checked by 5 analyses of the calibration mixture containing equal amounts of ah substances analyzed (500 ng in**  2  $\mu$ 1 of injected sample). The respective  $f_w$  values calculated from these analyses **did not exceed a 2% interval.** 

#### *Calibration of the column*

The column was calibrated by a mixture of equal amounts of cholesterol, **cholesteryl esters with 41,. 43,.45 and 47 carbon atoms and triglycerides with carbon numbers 48, 50, 52, 54 and 60. The calibration mixture was prepared from stock solutions containing 1' mg/ml of individual substances in a mixture**  isooctane—chloroform  $80:20$  (v/v). After drying, the mixture was dissolved in the internal standard solution (cholesteryl butyrate  $200 \text{ ng}/\mu$ ) and diluted standards were prepared according to Table I. All solutions were injected in **duplicate by means of the N-4950 autosampler; the volume of the injected**  samples was  $2 \mu l$ .

#### **Mufhematical** *processing of* **the calibration** *curue*

The  $f_{\omega}$  plot vs. the amounts of the individual components serves as a basis for mathematical processing of the calibration curves, with special considera**tion of the analytical use in the non-linear region of this plot. As the plot of**  *fw vs. tie* **amount of the test substance is convex, a multilinear approximation of n-l linear sections was used in order to simplify the problem of the calibra**tion curve  $(n = 15, i.e. 14$  sections). In Practice, the scatter of the measured **values may cause oscillations of the calibration curve. To solve this problem the values measured were replaced mathematically by another set of** *values*  **which meets the condition of a minimum scatter between these two sets, to**gether with that of the convex character of the calculated plot. The  $f_{\nu}$  values **were CalcuIated from eqn. 1 [3] \_** 

**weight (5%)** 

 $f_w = \frac{1}{\text{area}(\%)}$ **area (%) (1)** 

The  $f_{\text{WF}}$  values can be calculated from the formula

$$
f_{\mathit{wr}} = \frac{f_{\mathit{w}}}{f_{\mathit{w}\mathit{is}}} \tag{2}
$$

where  $f_{\text{ufs}}$  is the weight correction factor for the internal standard. From eqn. 1 and 2 we get the formula for calculation of  $f_{\text{nn}}$  values for individual comwhen the contract of the contract of the contract of the contract of  $\overline{\mathcal{M}}$  is the contract of the contra

$$
f_{\mathbf{w}\mathbf{r}} = \frac{m_{\mathbf{i}}A_{\mathbf{i}\mathbf{s}}}{m_{\mathbf{i}\mathbf{s}}A_{\mathbf{i}}} \tag{3}
$$

where  $f_{wri}$ ,  $m_i$  and  $A_i$  are the weight correction factor, the weight and the peak area for component *i*, respectively. Analogously,  $m_{\tilde{B}}$  and  $A_{\tilde{B}}$  are the weight and the peak area for the internal standard. atus sa ng ang ang maing

Indicating the measured peak areas as  $x_i$  and the corresponding calculated  $f_{wr}$  values as  $y_i$ , the problem can be formulated as follows: for the set of points  $y_i$ , a corresponding set of points  $y'_i$  should be found under the following conditions: The value of the expression  $z = (y_i - y'_i)^2$  is minimal and the system is described by  $n-2$  inequalities and a granite participating the state of par-

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$$
\frac{y_{i+1}-y_i}{\frac{y_{i+2}-y_{i+1}}{x_{i+1}-x_i}} \leq \frac{y_{i+2}-y_{i+1}}{x_{i+2}-x_{i+1}}
$$

By meeting these conditions, the convex shape of the calculated plot is ensured. From the mathematical point of view, it is necessary to find the extreme of a function, under certain limitations. In this case the minimum quadratic function with n variables should be found with simultaneous validity of  $n-2$  linear limitations. In practice, the problem can be solved by application of the saddle point theorem proved in 1951 by Kuhn and Tucker [7]. A modified algorithm

### **TABLE I**

#### PREPARATION OF DILUTED STANDARDS



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**of the simplex linear-programming method, described first by Wolfe 181, was used.** 

The programme output is in the form of a magnetic tape containing the **coordinates of the optimized points-and- the slopes of their conkcting Lines, as well as a printed table which renders possible the 'control-of the scatter of**  the measured points. The output further contains a table of the  $f_{\mu\nu}$  values for all the substances analyzed, which permits a check of the plot of  $f_{wr}$  vs. the **peak area (or alternatively peak height) and manual calculation if necessary. For triglycerides with carbon number of 56 and 58, the computerized**  calibration based on the linear interpolation of  $f_{\mu\nu}$  values between C-54 and C-**60 is performed.** 

## *Evaluation of the analyses of biological samples*

**For the quantitative analyses of biological samples, au internal standard method was used. For calculation of the results, a special programme was prepared, based on the following formula** 

$$
mg/dl_i = C \cdot f_{uri} \cdot A_i; \qquad C = \frac{10^{-4} V_t m_{is}}{V_p V_a A_{is}}
$$
 (4)

where C is the constant for each sample analyzed,  $V_t$  ( $\mu$ I) is the volume of **internal standard solution, which was used for dissolving the sample before**  analysis,  $V_p$  (ml) is the corresponding volume of plasma,  $V_a$  ( $\mu$ 1) is the volume of injected sample and  $m_{is}$  (ng) is the amount of internal standard in the in**jected sample.** 

#### **RESULTS AND DISCUSSION**

As observed by other authors, the  $f_w$  value depends on the amount and molecular weight of the substance analyzed  $[2, 5, 9]$ . However, different  $f_{\omega}$ **values have been published for the same substances [4; IO] obtained under comparable conditions, but with different colunins and apparatus. According**  to our present knowledge, a loss of the substances separated is caused by pyrolytic decomposition and irreversible sorption. It is also known that under strictly constant conditions,  $f_w$  values are highly reproducible  $[2]$ . From the chromatographic aspect, the  $f_w$  value increases with increased loading with stationary phase and with increased column length [3, 11]. The percentage of loading with stationary phase also influences the efficiency of the column. It is known from the literature, that columns with a higher percentage of loading need a longer time for stabilization than those with a lower one. A similar effect is also exerted by the column length. With such columns, relatively high  $f_w$  values are reached even after stabilization. For preparation of **the. column with the-highest recovery of the test substances, lower loading with the stationary phase is necessary. The support should be highly inert. Chromo**sorb W **HP**, Chromosorb 750, Gas-Chrom Q and Supelcoport coated with 1-**3%. OV-1 or.JXR were tested. Best rcssts wereob&ined with @aas column 0.5 m.**  $\times$  **2.0** mm I.D., packed with 1% OV-1 on Gas-Chrom Q (100-120 mesh).

#### **TABLE II**

## REPRODUCIBILITY OF THE GC DETERMINATION OF THE STANDARD MIXTURE

Each value represents average of three analyses: 2 ul of the standard mixture containing 100 ng/µl of each component were injected. The states for the state states and if the last





Fig. 1. The plot of  $f_{\mu r}$  vs. amount of the substance analyzed, for main components of the lipid profile.  $27$  = Free cholesterol;  $45$  = cholesteryl stearate;  $52$  = 1.3-stearate-2-palmitate.

Fig. 2. The plot of  $f_{wr}$  vs. amount of the substance analyzed, for triglycerides with different molecular weights.  $48$  = Tripalmitin;  $54$  = tristearin;  $60$  = triarachidin. 가격 끝 위

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对话语的话 法财富法状态 拖绳 in Japan the special infrastructure of the problems. The efficiency of the column, expressed as  $\Delta$  C<sub>48</sub>-54 [3], was sufficient. The column needed a relatively short time for stabilization. The series of the series of

The reproducibility of the calibration data plays an important role in the lipid analysis when measured in the region where  $f_{in}$  is independent of the amount of substance analyzed. This requirement is much more important in the non-linear region. The repeatability of the calibration data with time should also be checked as an important criterion of the reproducibility. Table II shows the results for repeated analyses of the calibration mixture (200 ng of each component) obtained with the column mentioned above. Similarly, the reproducibility of calibration was tested over the whole concentration range. It was confirmed that the quantitative lipid analysis in the non-linear



Fig. 4. Gas chromatograms of: (A) normal and (B) hyperlipidemic plasma neutral lipid.  $27$  = Free cholesterol;  $31$  = cholesteryl butyrate (internal standard);  $33$  = cholesteryl benzoate (standard for laboratory control);  $41 - 47$  = cholesteryl esters;  $48 - 58$  = triglycerides; sample volume =  $2 \mu l$ ; solvent = isooctane chloroform (80:20,  $\nabla/\nabla$ ); sensitivity = 1/64; chart speed  $=5$  mm/min; other analytical conditions are given in the text.

range of  $f_{\omega}$  values is sufficiently precise. The method described differs from that of the internal standard in that the  $f_w$  value is obtained from the  $f_w$  vs. peak-area calibration plot for the test substance. Figs. 1 and 2 shows these calibration plots for some individual substances.

As shown in Figs. 1 and 2, the higher the molecular weight of the substance, the more curved is the calibration plot. Furthermore, the parameters discussed previously, i.e. the quality of the column and the operating conditions also influence the shape of the calibration plot. It is known that the shape of the plot of  $f_{\mu}$  vs. the carrier gas flow-rate is the reverse of that for the separation ability vs. the flow-rate  $[3]$ . In our investigation, these conclusions were confirmed. Fig. 3 shows the optimization of the carrier gas flow-rate for the column described.



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#### **TABLE IV**

#### **CALIBRATION OF THE COLUMN No. 33**

 $\label{eq:1} \mathcal{L}^{\mathcal{A}} \left( \mathcal{L}^{\mathcal{A}} \right) = \mathcal{L}^{\mathcal{A}} \left$ 

Date: 11-11-77.



\*Integrator output divided by 5000.

We chose computerized processing of the calibration data, because manual processing is very laborious and lacking in precision. The basic mathematical assumptions have been discussed above. In this way, 500  $f_w$  values were obtained as a function of the peak area of the test substance. The programme permits two versions of data processing. As confirmed by Gold and Mathew [6], both the peak heights and their areas can be used for the quantitative analysis of lipids. The  $f_{\nu}$  vs. peak height plot is very similar to that of  $f_{\nu}$  vs. peak area. As mentioned above, mathematical processing permits the control of the accuracy of the calibration data, using a special optimization programme.

Table III shows an example of computerized calibration data for one compound; from this table the differences between the measured and optimized values and also between duplicate analyses of the same sample and the slopes of the individual linear regions are apparent. This method of data processing permits rapid control of the calibration accuracy. The difference between duplicate measurements of the same sample should not exceed 5% in relation to the mean  $f_{\mu r}$  value for each component. Analogously, this type of computer output serves as a rapid check of recalibration. As can be seen from Table II, the interval of recalibration is longer than 200 analyses. In practice, when the chromatograph is used daily, it represents a time interval of about  $1-2$  months. The calibration proper is automatic and takes 32 h. The calibration data are

recorded on a magnetic tape and used throughout the period of calibration validity for the evaluation of biological samples. The calibration is repeated if any of the substances display a deviation of  $f_{ur}$  values greater than 5% in any part of the plot. The problem of calibration stability will be discussed in the following part of this paper. s.

During the column life, the  $f_{\mu\nu}$  values decrease in the lower concentration range, especially in the case of triglycerides with higher molecular weight. These changes are smaller when the column is properly stabilized. Table IV illustrates part of the output of tabulated  $f_{wr}$  data in relation to the peak area; similar data were obtained for peak heights. Table V shows an example of the calculation for a biological sample. Gas chromatograms of normal and hyperlipidemic plasma neutral lipid are given in Fig. 4.

The proposed method permits on-line as well as off-line coupling with a computer. From economic aspects, the off-line coupling is more effective; it employs a conversion programme, using punched or magnetic tapes containing the input data in the ASC 11 code.

The method also permits the inclusion in the computerized programme of some additional parameters suitable for clinical purposes (e.g. various sums or ratios of measured values).

#### **TABLE V**

#### **CALCULATION OF THE LIPID PROFILE**





**\*Integrator output divided by 5000.** The concentration of the first of the first policies with

\*\*Content of cholesterol in individual ester fractions. And the set of the state of the state of

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